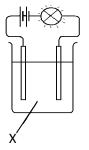
- (05) (A) The life process that gives rise to a new generation from an existing generation is referred to as reproduction.
 - 1. Name one plant doing vegetative propagation by following plant parts.
 - i. Roots
 - ii. Runners
 - iii. Rhizome
 - 2. Write two adaptations shown by plants to avoid self pollination and promote cross pollination.
 - 3. i. Name 2 glands associated with the male reproductive system.
 - ii. Name the follicle stimulating hormone.
 - iii. Write a function of epididymis of male reproductive system.
 - iv. Where does fertilization of ova takes place in the female reproductive system?
 - (B) Photosynthesis produces food in green plants.
 - i. What are the two external factors affect for the photosynthesis process?
 - ii. Following are several steps followed in an experiment which is done to show the sunlight is essential for photosynthesis. Complete the steps.
 - a. Pick a plant leaf which is exposed to sunlight.
 - b.
 - c. Warm in alcohol.
 - d.
 - e.
 - f. Obtain a dark purple colour.
 - (C) 1. The dominant gene for Hemophilia is H and recessive gene is h. When a carrier female is married to a healthy male, how does the inheritance of Hemophilia happen? Show in a genetic diagram.
 - 2. Genes of which plant inserted into the paddy to produce golden rice?
 - 3. Write one common inherited characteristic among human.
- (06) (A) Iron is extracted from the iron ore mined from the soil.
 - 1. What is the instrument used for this process?
 - 2. Following is one of the reactions happens inside it. (Fe = 56, O = 16)

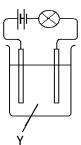
$$Fe_2O_{3 (s)} + 3 CO_{(g)} \longrightarrow 2 Fe_{(l)} + 3 CO_{2 (g)}$$

- i. Find the molar mass of Hematite.
- ii. What is the mass of Fe₂O₃ to be reduced to obtain 112 g of Fe?
- (B) Given below A and B set ups are prepared in the laboratory to identify types of bonds.

bulb is lighted

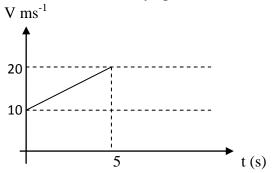
bulb is not lighted





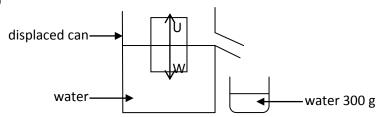
1. Write the types of bonds in X and Y separately.

- 2. Write 2 compounds used for X and Y separately.
- 3. Write 2 physical properties in compound X.
- (C) A, B, C, D, E, F, G and H are 8 consecutive elements that belong to second and third periods of the periodic table. B is a noble gas.
 - 1. Which element has the highest electro negativity?
 - 2. Which element has the highest first ionization energy?
 - 3. What is the formula of the compound made by C and A?
 - 4. Which element makes a strong basic oxide?
- (D) Salt is extracted by collecting sea water in tanks.
 - 1. What is the salt deposited in the first tank?
 - 2. What is the concentration of mother solution when NaCl deposition takes place?
- (07) (A) Following graph shows how an elevator with people moving at a constant velocity of 10 ms⁻¹, increased its velocity up to 20 ms⁻¹ within 5 s.



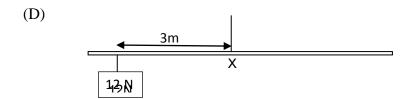
- 1. Find the acceleration of the elevator.
- 2. Find the displacement of the elevator during that time.
- 3. If the mass of the elevator with the people is 500 kg, what is the unbalanced force act on the elevator to move in acceleration? ($g = 10 \text{ ms}^{-2}$)
- (B) A space shuttle without people with the mass of 600 kg moved out from the earth's gravity to land on the moon.
 - 1. What is the weight of it at the earth?
 - 2. What is the reason for decreasing the weight of it at the moon?
 - 3. Write down the Newton's law of motion of the rocket when it take off from the earth.

(C)



Above is a set up arranged to demonstrate Archimedes Law. When the block of wood is immersed in the vessel X, 300 g of water got displaced. ($g = 10 \text{ ms}^{-2}$)

- 1. Write the Archimedes Law.
- 2. Find the up thrust acts on the block of wood.
- 3. Find the weight of the block of wood.

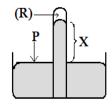


- 1. Find the anti clockwise moment of force.
- 2. Without removing 12 N weight and hanging 6 N weight at the right side it is expected to be balanced. How far the weight should hang from the point X?
- (08)(A) When more than one force is applied, the single force that gives the same result as that of all the contributing forces is known as the resultant force.
 - 1. Find the resultant force in following instances.



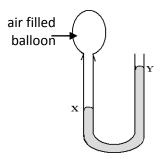
(B) Given below is a set up prepared with a liquid of 1500 kgm⁻³ density.

(P= atmospheric pressure= $1 \times 10^5 \text{ Nm}^{-2}$, $g = 10 \text{ ms}^{-2}$)



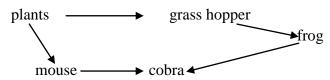
- 1. Find the height of the liquid column in the above given atmospheric pressure.
- 2. What has happened to set up when it takes 1 km above the sea level?
- 3. What is indicated by 'R' in the above set up?
- 4. Give 2 examples where atmospheric pressure is used.

5.



If this set up is exposed to sunlight about an hour what can you observe? Write 2 observations.

- (C) All living units show one or few living characteristics.
 - 1. What do you mean by growth?
 - 2. Given below is a food web.



- i. Name two organisms show heterotrophic mode of nutrition.
- ii. What is the energy source used by plants in their mode of nutrition?

- (D) 1. Write one function of each of the following organelles.
 - i. Mitochondria
 - ii. Endoplasmic reticulum
 - 2. Name the chemical compound made of plant cell wall.
 - 3. Name a place where meosis takes place in a living organism.
 - 4. What the type of division responsible for new characteristics?
- (09) (A) If the time taken by ultrasound waves transmitted by a ship which was in an exploration of debris of a ship to reach the detector again after reflection from the sea bottom is 3 s. (speed of sound in sea water is 1400 ms⁻¹)
 - 1. Find the depth of the sea.
 - 2. What is known as the Lithotripsy technology?
 - 3. Write 2 common characteristics of electro-magnetic waves.
 - 4. Write down the type of electro-magnetic waves used in following instances.
 - a) To check hidden symbols in currency notes.
 - b) To take photographs of internal organs of the human body.
 - 5. Write down the optical device used in following instances.
 - a) To diverge a parallel light beam.
 - b) To prepare a solar hearth.
 - (B) You are provided with following metals and answer using them only.

Na, Mg, Cu, Zn, Fe, Pb

- 1. Which metal is stored in paraffin oil?
- 2. Name 2 metals which can displace Cu from CuSO₄ solution.
- 3. Write 2 uses of Magnesium metal.
- 4. Name the metal which is not reacted with dilute acids.
- 5. Write observations when a small piece of Sodium is put into the water.
- 6. Write the method of extracting Sodium and Magnesium.